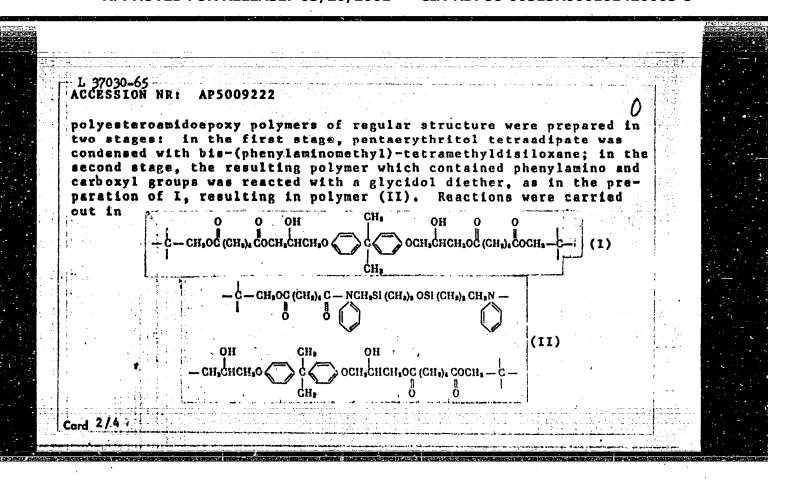
ANDRIANOV, K.A., "Tademik; DELAWARI, B.V.; Velaway, L.M.; EREMATEVERLY, L.A.

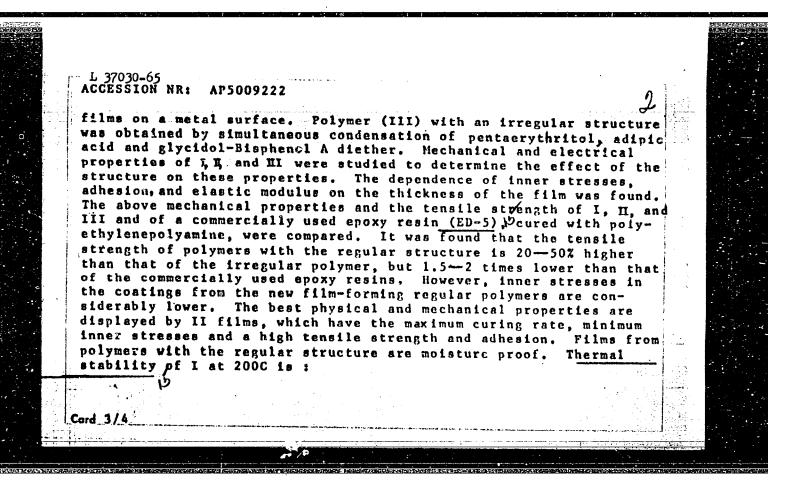
Synthesis and spectra of trimethylathyl-filenyl, shlero,-i-exa2,6-disilacyclohexanes, Dokl. AN S9Sk 160 m.J. 137 J-1316 F 165.

(EREA 18:2)

1. Institut elementoorganicheskikh soyedirenty IN SESS.

L 37030-65 EPA(s)-2/EWT(m)/EPF(c)/EWP(v)/EPR/EPA(w)-2/EWP(j)/TPc-4/Pab-10/Pr-4/Ps-4/Pt-10 WW/RM S/0020/65/161/001/0099/0102 ACCESSION NR: AP5009222 AUTHOR: Andrianov, K. A. (Academician); Yemel'yanov, V. N.; Sukhareva, L. A.; Smirnova, Yu. P.; Zubov, P. I. TITLE: Synthesis and physical and mechanical properties of films from polymers with regular structure SOURCE: AN SSSR. Doklady, v. 161, no. 1, 1965, 99-102 TOPIC TAGS: polymer, regular structure, regular structure polymer, epoxy polymer, polyesteroepoxy polymer, silicon containing polyesteroamidoepoxy polymer ABSTRACT: The purpose of the work was to synthesize and study polyesteroepoxyband silicon-containing polyesteroamidoepoxy polymers of a regular cyclonet structure, which could be used for coatings, letectric insulation or as binders for glass-reinforced plastics. Polyesteroepoxy polymers were obtained by reacting pentaerythritol tetraedipate with glycidol-hydroquinone or with glycidol-diphenylolpropane ("Bisphenol A") (See I below) diethers. Silicon-containing Card 1/4





L_37030-65 ACCESSION NR: AP5009222							
Test duration, hr Weight loss, %		100 0.95	500 3.60		1500 6.32		
Electric properties wer plastic, in which I was mulas, 3 graphs, and 2	used as	a binde	r. Or	ig. art.	hasi	4 for-	
ASSOCIATION: Institut							
ASSOCIATION: Institute nauk SSSR (Institute of Sciences, SSSR) SUBMITTED: 29Sep64				unds, A		<u>f</u> .	
nauk SSSR (Institute of Sciences, SSSR)		lemental	Сошро ¹	unds, A	ademy o	oc,	MT

EVT (m)/EPF(c)/EVP(1) Pc-4/Pr-4 ACCESSION NR: AP5006418 \$/0062/65/000/001/0167/0169 AUTHOR: Andrianov, K. A.; Sidorov, V. I.; Khananashvili, L. M.; Kuznetsova, N. TITLE: Reaction of the addition of methyldichloro- and dimethyl-chlorosilanes to vinyl-derivatives of organocyclosiloxanes SOURCE: AN SSSR. Izvestiya. Seriya khimicheskaya, no. 1, 1965, 167-169 TOPIC TAGS: silans, organosilicon compound, methyldichlerosilane, dimethylchloro-ABSTRACT: The reactions of the addition of methylchloro- and dimethylchlorosilanes to vinylheptamethyl- and divinylhexamethylcyclotetrasiloxanes in the presence of an H2PtCl6 catalyst occur relatively easily and with good yields of the desired products. The Si(CH₂)Cl₂ and Si(CH₃)₂Cl groups are joined to the 8-carbon atom of the vinyl group of cyclosiloxane. Five new compounds were synthesized. "The authors express their deep gratitude to M. G. Zaytseva for taking the infrared absorption spectra." Orig. art. has: 2 tables, 1 equation. Card 1/2

ACCESSION NR: AP5006418 ASSOCIATION: Moskovskiy	institut tonkov khimicheskov tecki	nnologii imeni M. V.	(
Lomonosova (Moscow Insti SUBMITTED: 22May64	tute of Fine Chemical Technology) ENCL: 00	SUB CODE: GC, OC	
NO REF SOV: 000	OTHER: 000		
	· `\		

<u>L 40978-65</u> EWT (m)/EPF(c)/EPR/EWP(j) Pc-4/Pr-4/Ps-4 RPL WW/RM ACCESSION NR: AP5006422 S/0062/65/000/001/0187/0189

AUTHOR: Andrianov, K. A.; Shapatin, A. S.; Ponomarev, V. V.

TITLE: Formation reactions and properties of aluminum salts of ethoxymethylphosphinic and diethylphosphoric acids

SOURCE: AN SSSR. Izvestiya. Seriya khimicheskaya, no. 1, 1965, 187-189

TOPIC TAGS: aluminum, aluminum compound, phosphonic acid, phosphoric acid, polymer

ABSTRACT: Aluminum diisopropoxy(ethoxymethylphosphinate), aluminum isopropoxy-bis(ethoxymethylphosphinate), and aluminum tris-(ethoxymethylphosphinate) were synthesized. The interaction of aluminum isopropylate/with triethylphosphate was studied
in molar ratios of 1:1, 1:2, and 1:3. The reaction of aluminum isopropylate with
ethyl ethers of methylphosphinic and phosphoric acids yields aluminum tris-(ethoxymethylphosphinate) and aluminum tris-(diethylphosphate). Aluminum isopropoxy-bis(ethoxymethylphosphinate), and aluminum tris-(ethoxymethylphosphinate) have a polymeric structure. Orig. art. has: 1 figure, 4 equations.

C-- 1/2

		1	
ASSOCIATION: Moskovskiy ind Lomonosova (Moscow Institute	stitut tonkoy khimicheskoy tek s of Fine Chemical Technology	khnologii im. M. V.	
SUBMITTED: 03Jun64	ENCL: 00	SUB CODE: GC, OC	
NO REF SOV: 002	OTHER: 000		
	en e	1 .	
•	•		
	A Commence of the Commence of		
•			
	·		

L 4	1575-65 Ession NF	R: AP50088	341	.gr is tre-suc = 1 cmm i i i		,	1	
sila meth	ane with hyldichlo, 5,7-tetuylcyclote	diethyldic prosilane pramethyl-1, etrasiloxan	chlorosilane. produced 1,3,5 ,5-divinylcycl ne, 1,5-dimeth methyl-3,5,7-t	Cohydrolysis 5,7-tetramethy Lotetrasiloxand myl-1,5-diviny trivinylsiloxan	of methylv: 1-1-vinylcyd 2-1-Methy 1-0,7-ulethy ne were pro	f methylvinyldi inyldichlorosil clotetrasiloxan l-1-vinyl-3,5,7 yicyciotetrasil duced by cohydr art. has: 1 ta	ane with e and -tri- loxane olysis of	
M.C.C.	,,							
ASS	OCIÁTION	: Noskovsk	kiy institut (stitute of Fir	tonkoy khimich ne Chemical Te	skoy tekhn chnology)	ologić im. N. V		
ASS	OCIÁTION onosova	: Noskovsk	kiy institut (stitute of Fir	tonkoy khimichene Chemical Te	eskoy tekhn chnology)	ologië im. N. V SUB CODE:		
ASS Lom SUB	OCIÁTION onosova	: Moskovsk (Moscow Ins	kiy institut (stitute of Fir	ne Chemical Te	00			
ASS Lom SUB	OCIÁTION CONOSOVA	: Moskovsk (Moscow Ins 02Jan64	kiy institut (stitute of Fir	e Chemical Te	onology)			
ASS Lom SUB	OCIÁTION CONOSOVA	: Moskovsk (Moscow Ins 02Jan64	kiy institut (stitute of Fir	e Chemical Te	onology)			
ASS Lom SUB	OCIÁTION CONOSOVA	: Moskovsk (Moscow Ins 02Jan64	kly institut (stitute of Fir	e Chemical Te	onology)			

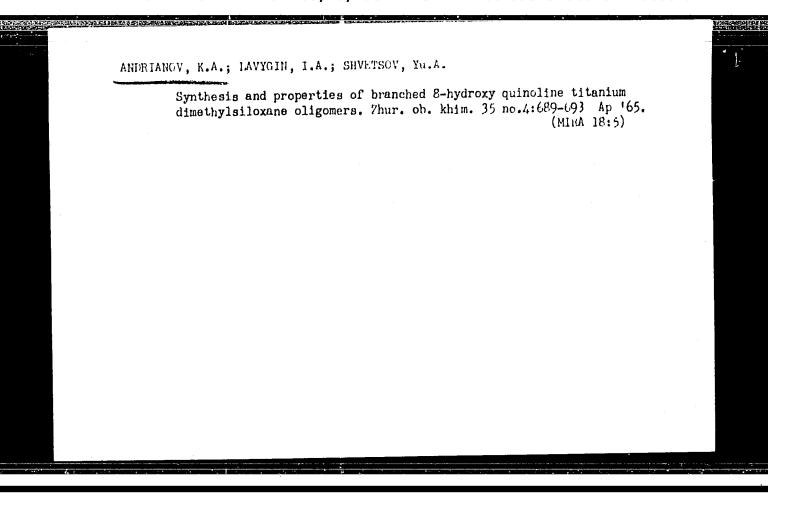
EWT(m)/EPF(c)/EPR/EWP(4)/T Pc_4/Pr-4/Ps-4 ACCESSION NR: AP5012450 UR/0062/65/000/004/0660/0665 546.287 AUTHORS: Fromberg, M. B.; Petrashko, Yu. K.; Vozhova, V. D.; Andrianov, TITLE: Double decomposition of alkyl(aryl)trisodium oxysilanes and methylphenyl dichlorosilane dichlorosilane SOURCE: AN SSSR. Izvestiya. Seriya khimicheskaya, no. 4, 1965, 660-665 TOPIC TAGS: silane, IR absorption spectrum, polymerization, polycondensation, sodium compound ABSTRACT: The double decomposition of trisodium salts of alkyl(aryl) silantriols and methylphenyl dichlorosilane was studied. In order to use the reaction for obtaining trifunctional splitting of oligomers with functional groups at the ends of the branches, the synthesis was carried out with 1 mole of alkyl(aryl) trisedium oxysilane for 3 moles of methylphenyl dichlorosilane. Sodium salts (obtained by treating alkyl(aryl)polysiloxanes with an alcohol solution of caustic soda) were used. The double decomposition reaction was carried out below 400 with gradual introduction into a solution of methylphenyl dichlorosilane of a suspension of the trisodium salt in toluene. Analysis of the resulting products Card 1/3

L 54445-65

ACCESSION NR: AP5012450

shows them to contain but an insignificant amount of functional groups. The chlorine content was but 0.1% as against an expected 17.17%, on the assumption of the course the reaction would follow. Only traces of the hydroxyl group were detected after treatment with water. These data indicate that the double decomposition does not follow the expected pattern, but that it is apparently accompanied by hydrolytic processes that lead to the formation of cyclic compounds of complex structure. This view is supported by the presence of crystallization water in alkyl(aryl) silantriols. For the double decomposition reactions, sodium salts of methyl, ethyl, and phenyl silantriols were used. These yielded 1,7dimethyl-3,5,9,11,14,16-hexamethylhexaphenyl bicyclo (5,5,5) octasiloxane; 1,7-diethyl-3,5,9,11,14,16-hexamethylhexaphenyl bicyclo (5,5,5) octasiloxane; and 1,7-dipheny1-3,5,9,11,16-hexamethylhexaphenyl bicycle (5,5,5) octasiloxane. These compounds are lew-viscosity liquids, soluble in benzene, toluene, and carbon tetrachloride, and insoluble in ethyl and methyl alcohols. The composition, structure, and properties of the compounds are tabulated. Infrared spectra of all compounds exhibit an absorption band in the 1080-1090 cm-1 region, corresponding to vibration of the Si-O bond in eight-member rings. No characteristic bond for Si-OH was detected. Supplementary experiments on catalytic polymerization and thermal polycondensation demonstrated that the compounds are polymerized by means of 1% NaOH at 80C and that thermal polycondensation, which was Card 2/3

	effected at changes in p support the figures, 1	properties or c view that the table, and 2 fo : Elektrotekhn	g long periods (up to 3 omposition of the synth compounds have cyclic s rmulas.	esized compounds. tructure. Orig. a	rt. has	2		THE PROPERTY IN THE PARTY IN TH
	SUBMITTED:	17Apr63	ENCL: 00	SUB C	ODE: 00	o, go		
	NO REF SOV:	003	OTHER: 002					
							1	0 40 45
•	- 1A		- 14		1.4.1 - 13 - 14			
	Cord 3/3		and the second s					



ANDRIANOV, K.A.; SIDOROV, V.I.; KHANANASHVILI. L.M.; KUZNETSCVA, N.V.

Reaction of the cohydrolysis of methylallyldichlorosilane with methyl- and ethyldichlorosilanes. Zhur. ob. khim. 35 no.4:698-700 Ap 165. (MIRA 18:5)

1. Moskovskiy institut tonkoy khimicheskoy tekhnologii imeni M.V. Lomonosova.

ZHINKIN, D.Ya.; MAL'NOVA, G.N.; POLONSKAYA, A.P.; ANDRIANOV, K.A.

Simultaneous hydrolytic condensation of trimethyl-, triethylchlorosilanes, and phenyltrichlorosilane. Zhur. ob. khim. 35 no.5:909-911 My '65. (MIRA 18:6)

ANDRIANOV, K.A., ZHDANOV, A.A., KASHUTINA, E.A.

Synthesis and study of the properties of polydimethylsiloxanes containing carboxyl groups in organic end radicals. Zhur. ob. khim. 35 no.6:1037-1040 Je '65. (MIRA 18:6)

ANDRIANOV, K.A.; IZMAYLOV, B.A.

Reaction of higher alkylchlorosilanes with ethyl alcohol. Zhur. ob. khim. 35 no.6:1041-1044 Je '65. (MIRA 18:6)

ACCESSION NR: AP5007659 S/0020/65/160/006/1307/1310 AUTHORS: Andrianov, K. A. (Academician); Delazari, N. V.; Volkova, L. Ne;	
TITLE: Synthesis and spectra of trimethylalkyl-(phenyl, chlor)-1-oxa-2,6-	
SOURCE: AN SSSR. Doklady, v. 160, no. 6, 1965, 1307-1310	
TOPIC TAGS: cyclohexane, IR absorption spectrum, spectrophotometer/VIKS M 3 spectrophotometer, IKS 14 spectrophotometer	
ABSTRACT: The authors have produced new trimethylalkyl-(phenyl, chlor)-1-oxa-2,6 disilacyolohexanes, with a yield of 60-80%, during hydrolysis of bis(alkylchlorsilyl) propanes by an aqueous solution of caustic potash. On heating an ether solution of bicyclic compound was obtained according to	
bicyclic compound was obtained according to Ch ₁ (CH ₂)Si(CH ₂)Si(CH ₂) ₃ Ci N ₂ HCO. CH ₂ CH ₃ CH ₄ CH ₄ CH ₄ CH ₅ CH ₆ CH ₆ CH ₆ CH ₇ CH ₈ CH	
(CH _a), Si Si — O — Si — Si — (CH _a).	

1	: AP5007659		
Bis-(alkyloh)	lorsilyl) propanes were obtained according to reactions	(a), (b), and	
	CH.(B)CISICHICH = CH./	(a) (b)	
	CH ₉ (R)CISICH ₂ CH = CH ₉ + HSi(CH ₉) ₂ CI H ₉ PtCl ₄	(c)	
on two spectr IKS-14 with a	of the newly synthesized substances are given in a tarobtained and compared with other compounds. These spec ophotometers: a VIKS M-3 with an NaCl prism (700-1500 cm KBr prism (400-700 cm 1). The spectra are illustrated. Orig. art. has: 1 table and 1 figure.	tra were studied	T.
ASSOCIATION:	Institut elementoorganicheskikh soyedineniy, Akademii r Hetero-Organic Compounds, Academy of Sciences SSSR)	nauk SSSR	
/ OI			
SUBMITTED: 20	50ot64 ENCL: 01 SUB Const.	∞, op	

 L 49286-65 EPF(c)/EWP(j)/EWT(m)/T/EWP(b)/EWP(t) Pc-4/Pr-4 IJP(c) JD/RM ACCESSION NR: AP5013754 UR/0020/65/162/002/0335/0338 AUTHOR: Zhdanov, A. A.; Andrianov, K. A. (Academician); Odinete, V. A.; Karpova, I. V. TITLE: Synthesis and polymerization of cyclotetrasiloxanes which contain heterocyclic radicals with a silicon atom SOURCE: AN SSSR. Doklady, v. 162, no. 2, 1965, 335-338	
TOPIC TAGS: organosilicon compound, cyclosiloxane, silahexyl substituted cyclotetra- siloxane, organosilicon compound polymerization	
ABSTRACT: An attempt was made to synthesize and polymerize cyclosiloxanes fringed with heterocyclic groups in which silicon atoms belonged simultaneously to the heterocyclic groups and to siloxane closed chains. No literature data were available on the compounds in question. To synthesize silacyclohexyl derivatives of cyclosiloxanes, 1,1-dichloro-3,4-benzo-1-silacyclohexane was hydrolyzed either alone or with dimethyldichlorosilane in motar ratios from 3:1 to 1:3. Mono-, di-, tri-, and tetra-(3,4-benzo-1-silahexyl)-cyclotetrasiloxanes were obtained. The properties of these compounds are tabulated in the original. The compounds obtained were polymerized in the presence of an alkaline catalyst, either 0.3% KOH or 0.3% tetra-	
Card 1/2	

n 49286-65

ACCESSION NR: AP5013754

methylammonium hydroxide, at various temperatures. Concentrated H₂SO₄ was ineffective as a catalyst. Formation of an insoluble product was observed in all cases of polymerization and was ascribed to the cross-linking effect of the alkaline agent, which either splits off organic radicals or opens the silahexyl rings. It was found that tetramethylammonium hydroxide is more effective as a polymerization catalyst than KOH; the reactivity of the cyclosiloxanes studied decreases with an increase in the number of heterocyclic radicals in the molecule. Orig. art. has: 2 formulas, label and 3 graphs.

ASSOCIATION: Institut elementoorganicheskikh soyedineniy Akademii nauk SSSR (Institute of Heteroorganic Compounds, Academy of Sciences, SSSR)

SUBMITTED: 11Jan65

ENCL: 00

SUB CODE: OC, GC

NO REF SOV: 004

OTHER: 003

ATD PRESS: 400A

Card 2/2

L 5h987-65 EWT(m)/EPF(c)/EPR/EWP(j)/T/ETC(m) Pc-1

Pc-l/Pr-l/Ps-l WW/RM UR/0363/65/001/003/0294/0300 541.6:542.9

ACCESSION NR: AP5011919

SSION NK: When I at a

AUTHOR: Andrianov, K. A.; Kurasheva, N. A.; Manucharova, I. F.; Berliner, Ye. MB

TITLE: Polymers with inorganic molecule chains, polytitanodimethylsiloxanes

SOURCE: AN SSSR. Izvestiya. Neorganicheskiye materialy, v. 1, no. 3, 1965, 294-300,

TOPIC TAGS: polymer, inorganic polymer, titanium organic compound, dimethylsiloxane rubber, polymethylsiloxane

ABSTRACT: Kinetics of cross-linking of polytitanodimethylsiloxanes, (product having a reticular structure), was studied using oligomers of general formula

TI $\left[\begin{pmatrix} CH_s \\ OSI_{--} \\ CH_s \end{pmatrix}\right]_n$ OH

as a model compound (where n is 2, 3, 23, 34, 52, and 128). The oligomer condensation reactions and cross-linking process took place simultaneously. The yield of the reticular product (gel) is directly proportional to the reaction duration. The

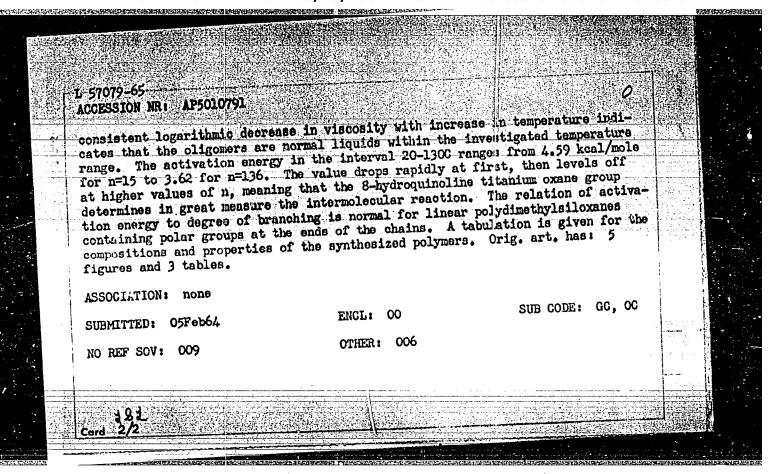
Card 1/2

L 54987-65 ACCESSION NR: AP5011919 rate of the cross-linking reaction is inversely proportional to In general, the polytitanodimethylsiloxanes are slightly more that the corresponding polymethylsiloxanes. Orig. art. has: 3 table ASSOCIATION: Institut elementoorganicheskikh soyedineniy Akadem (Institute of Organoelemental Compounds, Academy of Sciences, SSR (Institute of Sciences) shockey i neorganicheskoy khimii Akademii nauk SSSR (Institute of Organoelemental Compounds).	nii nauk SSSR	
SUBMITTED: 04Dec64 NO REF SOV: 005 ENCL: 00 OTHER: 000	SOB CODE:	•
Card 2/2		

ACCESSION NRT APSOLISZO	√P(j) — Po-L/Pr-4 — RM UR/0363/6	5/001/003/0301/0306	
	546.824:51	4-126	
AUTHOR: Andrianov, K. A.; Ki	iznetaova, I. K.; Smirnov, Yu. F	. 28	
TITLE: Reactions of titanius	s tetra-normal butoxide with pho	Osnbonia edide	
	Neorganicheskiye materialy, v.	and the control of the first state of the control o	
	le, titanium tetrabutoxide, phos		
	molecular weight of 1500 to 200 houghates were synthesized from		
tetra-monomethylphosphate tre	onic- and a-phenylvinylphosphon	ic acids. The titanium	
methylphosphate. Also two mo	d and a product of condensation it from each mole of starting t les a-phenylvinylphosphonic aci	itanium tetra-mono-	
	lvinylphosphate. The product o emical formula is C2H8O7P2Ti.		
Card 1/2			

	L 51699-65 ACCESSION NRI : AP5011920	
	tetra-u-phenylvinylphosphata with an excess of triethylbutoxysilanes and titanium	
. 1	tetra-normal butoxide leads to a substitution of hydrogen in the phosphonyl group	
	by triethylsiloxy- and titanium tri-normalbutoxy groups. The titinium tetra-mono- methylphosphate reacts with triethylbutoxysilane with formation of butyl alcohol	
	and di-triethylsiloxy-ester of methylphosphonic acid, accompanied by formation of	
	an insoluble product of the formula: C2H6O6P2Ti. In general, two alkylphosphonyl groups split off readily from the titanium tetra-alkylphosphates. It is concluded	
dichi 🕽	KLOADS - SDITT - OLI "IRAGITA" LIOM: CHA "ET CHILAM - CA CLA - ATVATDHOSHUA CR - H- II- "IR- CONCINARA	
<i>i,</i> ji	that only two phosphonyl groups can coordinate with a titanium atom. Orig. art.	
11	that only two phosphonyl groups can coordinate with a titanium atom. Orig. art	
1	that only two phosphonyl groups can coordinate with a titanium atom. Orig. art has: 1 figure and 4 formulas.	
	that only two phosphonyl groups can coordinate with a titanium atom. Orig. art.	
	that only two phosphonyl groups can coordinate with a titanium atom. Orig. art. has: I figure and 4 formulas. ASSOCIATION: Institut elementoorganichesikh soyadinaniy Akademii nauk SSSR (Institute of Organoelemental Compounds, Academy of Sciences, SSSR)	
	that only two phosphonyl groups can coordinate with a titanium atom. Orig. art. has: 1 figure and 4 formulas. ASSOCIATION: Institut elementoorganichesikh soyedineniy Akademii nauk SSSR (Institute of Organoelemental Compounds; Academy of Sciences, SSSR) SUBMITTED: 24Nov64 ENCL: 00 SUB CODE: 00, GC	
	that only two phosphonyl groups can coordinate with a titanium atom. Orig. art. has: I figure and 4 formulas. ASSOCIATION: Institut elementoorganichesikh soyadinaniy Akademii nauk SSSR (Institute of Organoelemental Compounds, Academy of Sciences, SSSR)	
	that only two phosphonyl groups can coordinate with a titanium atom. Orig. art. has: 1 figure and 4 formulas. ASSOCIATION: Institut elementoorganichesikh soyedineniy Akademii nauk SSSR (Institute of Organoelemental Compounds; Academy of Sciences, SSSR) SUBMITTED: 24Nov64 ENCL: 00 SUB CODE: 00, GC	
	that only two phosphonyl groups can coordinate with a titanium atom. Orig. art. has: 1 figure and 4 formulas. ASSOCIATION: Institut elementoorganichesikh soyedineniy Akademii nauk SSSR (Institute of Organoelemental Compounds; Academy of Sciences, SSSR) SUBMITTED: 24Nov64 ENCL: 00 SUB CODE: 00, GC	
	that only two phosphonyl groups can coordinate with a titanium atom. Orig. art. has: 1 figure and 4 formulas. ASSOCIATION: Institut elementoorganichesikh soyedineniy Akademii nauk SSSR (Institute of Organoelemental Compounds; Academy of Sciences, SSSR) SUBMITTED: 24Nov64 ENCL: 00 SUB CODE: 00, GC	

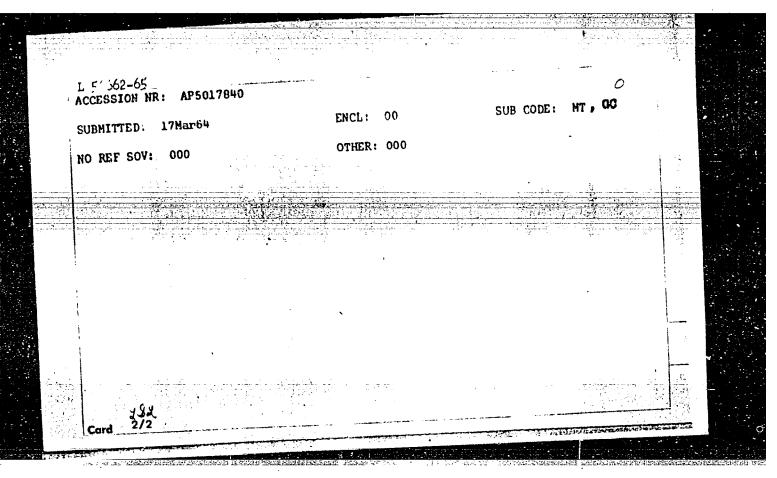
	IJP(c) JD/HA	:-
1 57079-65 EWI(m)/EPF(e)/EWP(j)/n) ser(t)/ser(b) Pc-4/Pr-4	
L 57079-65 EWILING AP5010791 ACC25 JON NR. AP5010791		. • -
	nvygin, I. A. Shvetsov, Yu. A.	
A CONTROL OF THE CONT	A 20 hydroxquinoline titarium	
TITL Synthesis and properti	les of branching 8-hydroxyquinoline titarium	
TITE 1 1 DXBUSS OF OTTROPIOT		
SOURCE: Zhurnal obshehey khir	mii, v. 35, no. 4, 1965, 689-69	
医三基氏性皮肤 自己 自然 化化学学 医黑色 医乳头 医乙二氏管 医化学性神经神经神经神经	MARKET TECHT IN SEA + and time to Depart of the Part	
TOPIC TAGS: polymer, organic glass transition temperature,	IR spectroscopy, viscosity	
ABSTRACT: The synthesis and	some properties of the liquid is viscous properties of the liquid is viscous tributation of the synthesis of (I) branching ends are described. The synthesis of (I)	
Siloxane of the ord to the	menching ends are described.	
was effected by condensation	branching ends are described. The synthem with alpha- of 8-hydroxyquinoline tributox; titanium with alpha- of 8-hydroxyquinoline tributox; titanium with alpha- of 8-hydroxyquinoline tributox; titanium with alpha- of 8-hydroxyquinoline tributox in which ydimethylsiloxane branching is 10, 15, (n) of the trimethylsiloxane branching is 10, 15,	
hydroxy-omega-or importantion	(n) of the trimethylelloxane utanomina =102 to	
-118C, and the refractive in	dex declines systematically with included a structure was studied by IR spectroscopy. A	Sec. Sec. Sec. Sec. Sec. Sec. Sec. Sec.
polymer12av1011.		
Cord 1/2		



UR/0020/65/161/004/0833/0835 ACCESSION NR: AP5010831 Wakarova, L. I.	
AUTHOR: Andrianov, K. A. (Academician); Makarova, L. I. AUTHOR: Synthesis of organosilicon multihydric ether alcohols TITLE: Synthesis of organosilicon multihydric ether alcohols SOURCE: AN SSSR. Doklady, v. 161, no. 4, 1965, 833-835	
TOPIC TAGS: organosilicon, alcohol, polyhydric alcohol, center alcohol, center alcohols containing B-oxyethoxymethyl groups ABSTRACT: Tri- and tetranydric ether alcohols containing B-oxyethoxymethyl groups Synthetic procedures and Synthetic procedures and Synthesized were: Acteristics are given in detail. Synthesized were: Acteristics are given in detail. Characteristics (Characteristics) - procedures and convergence of the convergence of t	
HS: (OC: Hs): € 3 (CHs): SCH: OCH: OCH: OCH: OCH: OCH: OCH: OCH: O	•
Cord 1/2	

equate 15 form formation of ethyl to the property of the formation of the second of th			3.25 (1.25 (
ACCESSION NR: AP5010831 In the case of tetraethoxysilane a corresponding tetrahydric ether alcohol, In the case of tetraethoxysilane a corresponding tetrahydric ether alcohol, It is postulated that in the intermediate SI[0Si(CH ₃) ₂ CH ₂ OCH ₂ CH ₂ OH ₃ H ₄ , was obtained. It is postulated that in the intermediate SI[0Si(CH ₃) ₂ CH ₂ OCH ₂ CH ₂ OH ₃ H ₄ , was obtained. It is postulated that in the intermediate SI[0Si(CH ₃) ₂ CH ₂ OCH ₂ CH ₂ OH ₃ H ₄ , was obtained. It is postulated in which is the intermediate of				
ACCESSION NR: AP5010831 In the case of tetraethoxysilane a corresponding tetrahydric ether alcohol, In the case of tetraethoxysilane a corresponding tetrahydric ether alcohol, Si[0Si(CH3)2CH2CH2CH2CH2CH2CH2, was obtained. It is postulated that in the intermediate Si[0Si(CH3)2CH2CH2CH2CH2CH2, was obtained. It is postulated that in the intermediate Stage of hydrolysis of 1,1-dimethoxysil-2,5-dioxycyclonexane a corresponding Stage of hydrolysis of 1,1-dimethoxysil-2,5-dioxycyclonexane a corresponding. The con- liance is formed which, in turn, enters into a condensation resident with thoxy- stage of hydrolysis of fermation of ethyl closed and a Si-dicit bending. The con- stage of hydrolysis of terminal condensation resident with thoxy- stage of hydrolysis of turn, enters into a condensation resident with thoxy- stage of hydrolysis of turn, enters into a condensation resident with thoxy- stage of hydrolysis of turn, enters into a condensation resident with thoxy- stage of hydrolysis of 1,1-dimethoxysil-2,5-dioxycyclonexane a corresponding The con- stage of hydrolysis of 1,1-dimethoxysil-2,5-dioxycyclonexane a corresponding The con- stage of hydrolysis of 1,1-dimethoxysil-2,5-dioxycyclonexane a corresponding The con- stage of hydrolysis of 1,1-dimethoxysil-2,5-dioxycyclonexane a corresponding The con- stage of hydrolysis of 1,1-dimethoxysil-2,5-dioxycyclonexane a corresponding The con- stage of hydrolysis of 1,1-dimethoxysil-2,5-dioxycyclonexane a corresponding The con- stage of hydrolysis of 1,1-dimethoxysil-2,5-dioxycyclonexane a corresponding The con- stage of hydrolysis of 1,1-dimethoxysil-2,5-dioxycyclonexane a corresponding The con- stage of hydrolysis of 1,1-dimethoxysil-2,5-dioxycyclonexane a corresponding The con- stage of hydrolysis of 1,1-dimethoxysil-2,5-dioxycyclonexane a corresponding The con- stage of hydrolysis of 1,1-dimethoxycyclonexane a corresponding The con- stage of hydrolysis of 1,1-dimethoxycyclonexane a corresponding The con- stage of hydrolysis of 1,1-dimethoxycyclonexane a corresponding Th	.t. 52266 - 65		C	
In the case of tetraethoxysilane a corresponding to the intermediate Si[OSi(CH ₃) ₂ CH ₂ OCH ₂ CH ₂ CH ₂ OH] ₄ , was obtained. It is profulate! that in the intermediate Si[OSi(CH ₃) ₂ CH ₂ OCH ₂ CH ₂ CH ₂ OH] ₄ , was obtained. It is profulate! that in the intermediate of hydrolysis of 1,1-dimethoxysil-2,5-dicxycyclonexane a corresponding stage of hydrolysis of 1,1-dimethoxysil-2,5-dicxycyclonexane in turn, enters into a condensation resulting with thoxysil-2, and the condensation of ethyl dicyclolated Silver and the condensation resulting formation of ethyl dicyclolated Silver and the condensation of ethyl dicyclolated Silver and the co	AP5010831			
stage of hydrolysis of 1,1-dimethoxysil-2,5-dimethoxysil-2,5-dimethoxysil-2,5-dimethoxysil-2,5-dimethoxysil-2,5-dimethoxysil-2,5-dimethoxysil-2,5-dimethoxysil-2,5-dimethoxysil-2,5-dimethoxysil-3,5-dimethoxysil-	one of tetraethoxysile	ane a corresponding tetrah as obtained. It is postul	ydric ether alcohol, ate! that in the intermedia wayane a correspondia;	ite
ASSOCIATION: none SUBMITTED: STIP CODE: OC. GC SUBMITTED: STIP CODE: OC. GC OTHER: OOS NO REF SOV: OO2	si[OSi(CH3)2cH2ocH2ocH2ocH2ocH2ocH2ocH2ocH2ocH2ocH2	methoxysil-2,5-mioxyc) arrn, enters into a condense on of ethyl miores in arrival arrival.	mion reading with cally con- ct-o-Si bonding. The con- cut to a conc. The pos-	•
NO REF SOV: 002	ASSOCIATION: none	ENCL: OO	SUB CODE: OC, OC	
NO REF SOV: 002	S. BMITTLD: C. JIPCD"	0.05		
Card 2/2?//		OTHER: DUS		
Card 2/2?np				
Card 2/2 ////			and the second s	
	Card 2/2 ///	and the same of th		

L 56662-65 EWT(m)/EPF(c)/EPR/EWP(j)/T Pc-4/Pr-4/Ps-4 WW/RM UR/0286/65/000/011/0078/0078 ACCESSION NR: AP5017840 678.84 3/	
AUTHOR: Andrianov, K. A.; Pakhomov, V. I.; Lapteva, N. Ye.	B.
SOURCE: Byulleten' izobreteniy i tovarnykh znakov, no. 22,	
ABSTRACT: This Author's Certificate introduces a method that the stability and improved mechanical characteristics are produced by carrying out the stability and improved mechanical characteristics are produced by carrying out the stability and improved mechanical characteristics are produced by carrying out the stability and improved mechanical characteristics are produced by carrying out the stability and improved mechanical characteristics are produced by carrying out the stability and improved mechanical characteristics are produced by carrying out the stability with oxyphenylpropylsilane and using aldehydes or aldehyde der	e management
 hydrolysis jointly with hardening catalyst. vatives as the hardening catalyst.	
vatives as the hardening catalyatt ASSOCIATION: Nauchno-issledovatel'skiy institut plasticheskikh mass (Scientific Research Institute of Plastic)	
Card 1/2	
COIO 1/2	हरमा दानसङ्ग्रहमा स्थानसङ्ग्रहमा स्थित । ०



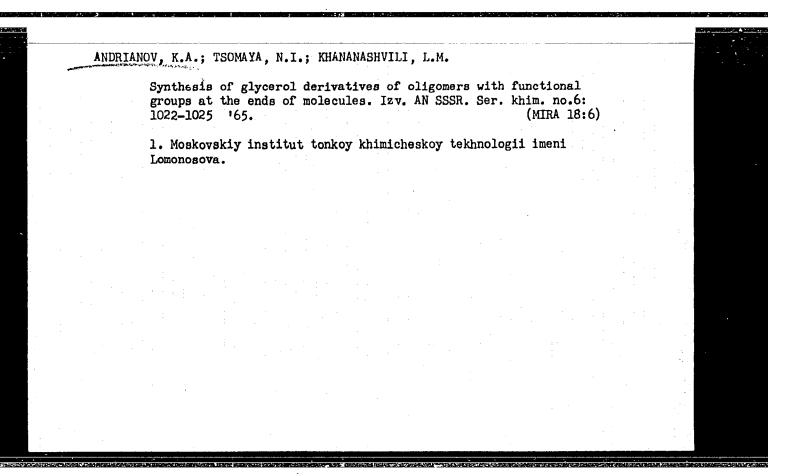
EWT(m)/EPF(c)/EWP(j)/T UR/0286/65/000/011/0080/0080 ACCESSION NR: AP5017848 678.84+678.643 AUTHOR: Prutkov, L. M.; Andrianov, K. A.; Polikanin, N. A.; Asnovich, E. Zi TITLE: A method for producing molding compounds. Class 39, No. 171577 SOURCE: Byulleten' izobreteniy i tovarnykh znakov, no. 11, 1965, 80 TOPIC TAGS: molding material, plastic, graft copolymer ABSTRACT to This Author's Certificate introduces a method for producing molding compounds based on a filler and binders -- synthetic resins. The useful properties of the molding compounds are improved by using binders consisting of graft copolymers of apoxy resin and a polyorganosiloxane which contains a secondary amino group in the side chain. 15 ASSOCIATION: none SUB CODE: MT. GC ENCL: SUBMITTED: 24Mar62 OTHER: 000 NO REF SOV: 000

ANDRIANOV, K.A.; KHANANASHVILI, L.M.; TELESHEVA, N.A.; TIKHONOV, V.S.

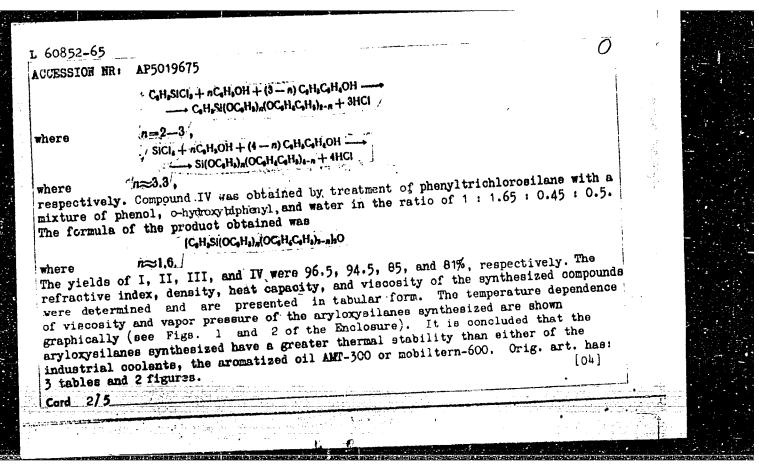
Reactions of dimethyloyolosilazanes with n-butyl alcohol and n-butyl borate. Izv. AN SSSR. Ser. khim. no.31446-449 '65.

(MIRA 18:5)

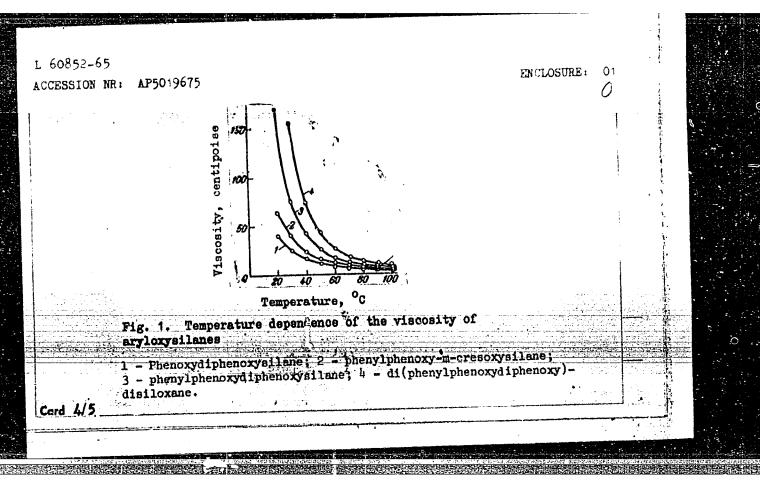
1. Institut tonkoy khimicheskoy tekhnologii im. M.V.Lomonosova.

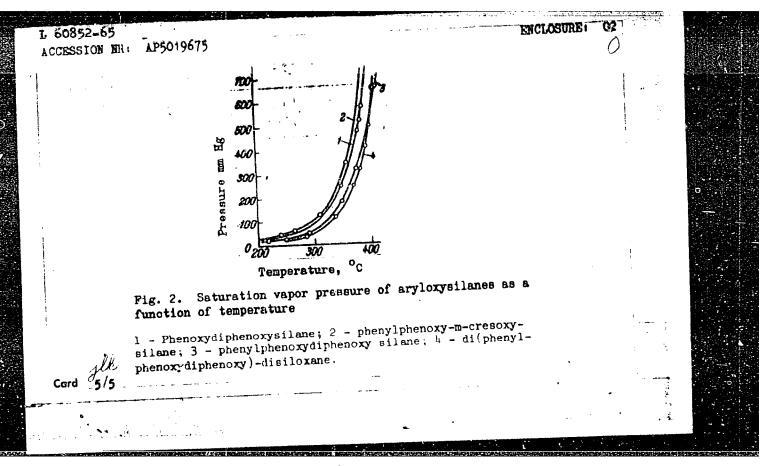


)		A STATE OF THE PARTY OF THE PAR					
L	60852-65 E	PA(s)-2/EWT(m)/EPF(c)/EW	P(1)/T WW/D		/0572		
. 1	ACCESSION NR:	AP5019675		UR/0064/65/000/008/05 662.987.9:547.11128			
		anov, K. A.; Ovchinnikov,	V. A.; Khanana	shvili, L. M.55,44	28 B	,	
						√ 5.	
		ilicon coolants cheskaya promyshlennost', r	10. 8, 1965, 57	0-572			
	MODIC MACS: O	ganosilicon compound, cool	lant, silane, s	iloxane/AMT 300 oil,	. •		
-				1.4	1		a a
	1 nw+0\\0	purpose of the investigat					
	phenylphenoxy-	Wich (T)	envinhenoxydip	nenoxy) disiloxane (1)	v). The		
	synthesis of I	xysilane (III), and di-(ph , II, and III was carried , CaHaSiCi, + nCaHaOH + (3 - n) CHaC		3 B0001 III0	į		
			-a + 3HCl			177 177 177	
	ayere	n≈1,8,					
	Card 1/5			and the second s			
	Mingration (1997)				-	<u>.</u>	
				0		45 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 -	يومش جهيم فيفري البير



ASSOCIATION: none SUBMITTED: 00 NO REF SOV: 000	ENCL: 02 OTHER: 005	SUB CODE: OC ATD PRESS:4063
Card 3/5		





FROMBERG, M.B.; PETRASHKO, Yu.K.; VOZHOVA, V.D.; ANDRIANOV, K.A.

Exchange decomposition reaction between alkyl (aryl) trisodiumhydroxysilanes and methylphenyldichlorosilane. Izv. AN SSSR. Ser. khim. no.4:660-665 '65. (MIRA 18:5)

1. Elektrotekhnicheskiy institut im. V.I.Lenina.

RPL JAJ/RM UR/0079/65/035/006/1037/10/10 EWT(m)/EPF(c)/EMP(j) Pc-4/Pr-4 L 61486-65 ACCESSION NR: AP5016407 546.287: 547.463 AUTHOR: Andrianov, K.A.; Zhdanov, A.A.; Kashutina, E.A. TITLE: Synthesis and study of the properties of polydimethylsiloxanes containing carboxyl groups in the terminal organic radicals SOURCE: Zhurnal obshchey khimii v. 35, no. 6, 1965, 1037-1040 TOPIC TAGS: organosilicon compound, heteroorganic polymer, polysiloxane synthesis, terminal carboxyl group, thermomechanical property, glass temperature ABSTRACT: In the catalytic polymerization of octamethylcyclotetrasiloxane in the presence of bis (2-carboxyethyl)tetramethyldisiloxane, linear polydimethylsiloxanes with carboxyl groups in the terminal organic radicals are formed: HOOCGH.CH.S Cord 1/2

L 61486-65

ACCESSION NR: AP5016407

$$- \frac{\text{GH}_{3}}{\text{CH}_{3}} = \begin{pmatrix} \text{CH}_{3} \\ \text{OS}_{1} - \\ \text{CH}_{3} \\ \text{CH}_{3} \end{pmatrix} = \begin{pmatrix} \text{CH}_{3} \\ \text{OS}_{1} - \text{CH}_{2} \text{CH}_{3} \text{COOH} \\ \text{CH}_{3} \\ \text{OS}_{1} - \text{CH}_{2} \text{CH}_{3} \end{pmatrix}$$

The average length of the oligomer formed is determined by the molar ratio of the reacting components. The physicochemical characteristics (including viscosity and activation energy of viscous flow) of the synthesized oligomers are tabulated. As the chain length increases, the molecular interaction declines because of a decrease in the concentration of the carboxyl groups per unit volume—and the properties of the oligomers increasingly resemble those of polydimethylsiloxanes with terminal trimethylsiloxy groups, since the influence of the terminal carboxyl groups regularly falls off. A study of the thermomechanical properties of the oligomers showed that they all have a glass-transition temperature between -80 and -110C under a 30 g load. IR spectra of all the oligomers were identical in character. The experimental procedure employed in synthesizing f(x) = f(x) = f(x) and f(x) = f(x) bis (2-carboxyethyldimethylsiloxy)polydimethylsiloxane (n = 10) is described. Similar compounds with n = 21, 28, 59, 73, and 106 were obtained. Orig. art. has:

ASSOCIATION: none

BUBMITTED: 05Feb64

NO REF SOV: 000

ENCL: 00

SUB CODE: OC

OTHER: 006

Card 2/2

ACCESSION NR: AP5011918	UR/0363/65/001/003/0289/0293
in a second seco	1 346.821:541.6 / Jul 36
AUTHOR: Andrianov, K. A.; Kuznetsova, I. K.	; Smirnov, Yu. N.
TITLE: Reaction of tetrabutyl titanate with	methyl-phosphonic acid esters
SOURCE: AN SSSR. Izvestiya. Neorganicheskiy 293	e materialy, v. 1, no. 3, 1965, 289-
TOPIC TAGS: titanate, organotitanium compou	nd, oligomer, polycondensation
ABSTRACT: Oligomers are prepared with chain tanium, oxygen and phosphorus by polycondens dialkyl esters of methyl-phosphonic acid. T	he reaction was carried out at 170-
-700°C without catalysts at initial componen condensation of n-butyl titanate with dibuty 1:1 ratio takes place with isolation of the	it ratios of 1:1, 1:2 and 2:1. Poly-
mers:	

L 65216-65

ACCESSION NR: AP5011918

$$n(n-C_{4}H_{9}O)_{4}Ti + n(C_{4}H_{9}O)_{2}-P-CH_{3} \rightarrow (2n-1)C_{4}H_{9}OC_{4}H_{9}$$

$$O$$

$$+C_{4}H_{9}-\begin{pmatrix} OC_{4}H_{9} & CH_{2} \\ -O-Ti-O-P- \\ | & | \\ OC_{4}H_{9} & O \end{pmatrix}_{n} -OC_{4}H_{9}$$
(A)

The authors studied the effect of temperature on the rate of polycondensation. The rate of the reaction was monitored by checking the amount of isolated dibutyl ester. It was found that an increase in temperature increases the completeness and speed of the reaction as well as the oligomer yield. An increase in the duration of isothermal nolding at 200°C during the reaction increases the completeness of the reaction and the titanium content in the reaction mixture. However, the relative viscosity of the oligomer solution during polycondensation increases very little (from i.11 to 1.56). The oligomer prepared by condensation at 200°C for 50 hours with subsequent removal of volatile products at 200°C and 1-0.1 mm Hg is a resinous dark yellow substance with a molecular weight of 4000 which la quite solu-

L 65216-05

ACCESSION NR: AP5011918

ble in alcohols, aromatic hydrocarbons and petroleum ether. The oligomer is easily hydrolyzed with the isolation of butyl alcohol and the formation of an insoluble product. This polymer is deformed at 100°C, but does not flow even at 400°C. Condensation of n-tetrabutyl titanate with dibutyl ester of methyl-phosphonic acid in a 1:2 ratio takes place with the formation of a monomer product:

However, reaction of these same initial components in 32:1 ratio takes place according to scheme (A) with the formation of a polymer product and the excess (1 mol) n-tetrabutyl titanate is returned from the reaction. Ultimate analysis and examination of the properties of this polymer product indicate that it is close in tructure to the product formed from an initial component ratio of 1:1. The experimental work is described in detail. Orig. art. has: 4 figures, 2formulas, 2 tables.

Card 3/4

L 65216-65

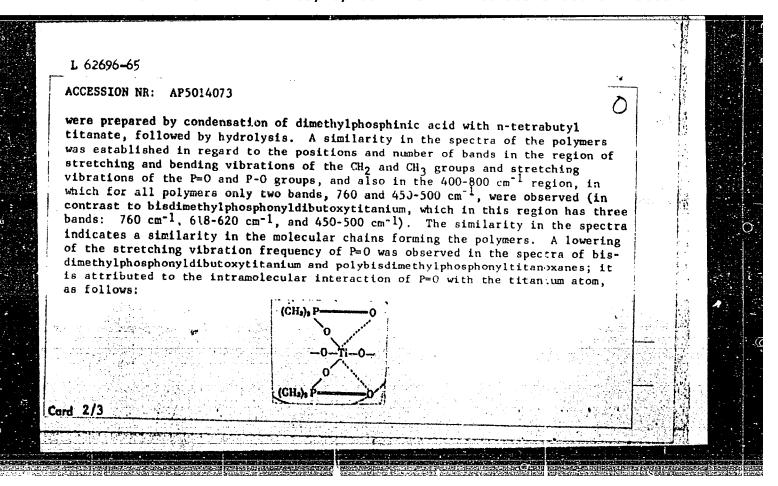
ACCESSION NR: AP5011918

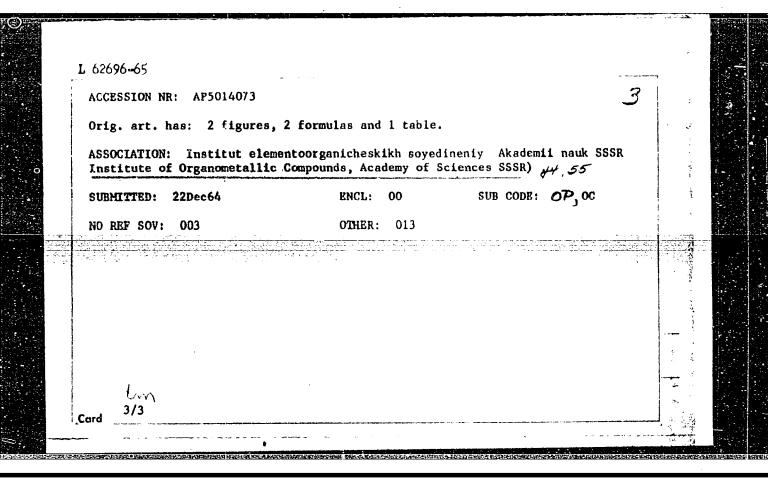
ASSOCIATION: Institut elementoorganicheskikh soyedineniy Akademii nauk SSSR
(Institute of Hetero-organic Compounds, Academy of Sciences SSSR), C.

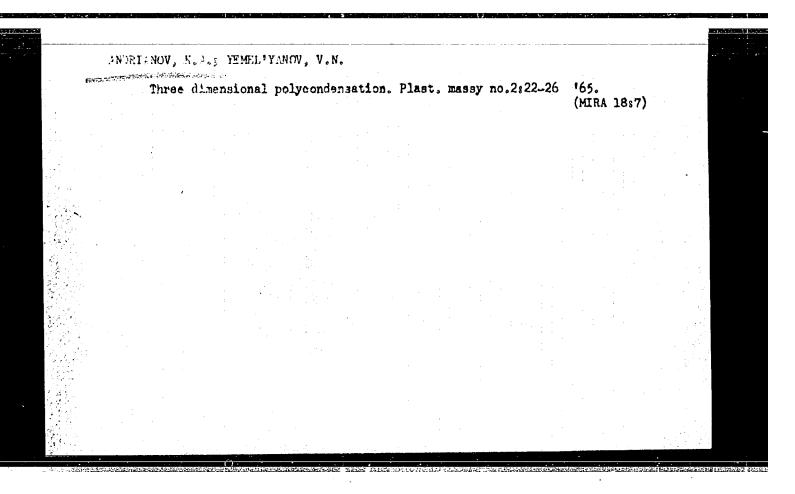
SJBMITTED: 200ct64 ENCL: 00 SUB CODE: GC, 0C

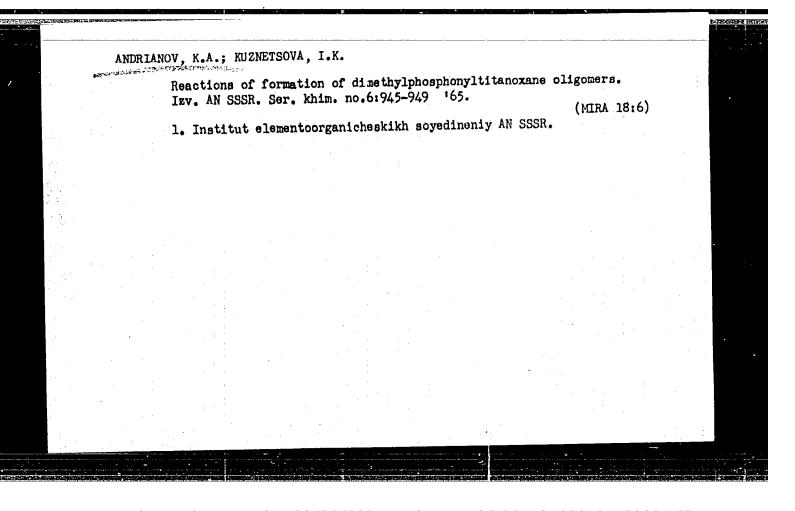
NO REF SOV: 001 OTHER: 000

ACCESSION NR:		UR/0363/65/001/0 541.6:543.422.4	20	,
AUTHOR: Andria	nov, K. A.; Gashnikova, N	بير 55 البر I. P.; Kuznetsova, I. K	44,56 J.J.	
	s with inorganic molecula		Acres 4	
SOURCE: AN SSS 460-463	R. Izvestiya. Neorganich	eskiye materialy, v. l	, no. 4, 1965,	
	frared spectrum, titanium	containing polymer i	norganic polymer,	Control of the second
	ared absorption spectra o			American Control
	C ₄ H ₅ -	-OC4He		









			0
	L 60045-65 EWT (m)/Lif(c)/EdF(J) Pc-4/FT-4 JAJ/AM ACCESSION NR: AP5018036 UR/0191/65/000/007/0023/0026 678.84		
	AUTHOR: Andrianov, K. A.; Yemel'yanov, V. N.; Raygorodskiy, I. M.		
	TITLE: Three-dimensional condensation of silicon-containing diamines with tetrafunctional ester, acids	্ধাৰ্ম ক্ৰ	
	SOURCE: Plasticheskiye massy, no. 7, 1965, 23-26		
	TOPIC TAGS: organosilicon compound, ester acid, condensation reaction, poly-	# 18 mm 1 mm	
	amide, dismine ABSTRACT: The article presents the results of a study of three-dimensional con-	() () () ()	
	densation of tetrafunctional ester acids with organistrical	*	
	GCH,OC(CH,),COOH,+ 2HNCH,(SIO),SICH,NH -HO		
	CH, CH,		
	CH, CH,		
	-C-CH ₂ OC(CH ₂) _n C-NCH ₂ (SIO) _x -SiCH ₂ N-C(CH ₂) _n COCH ₂ -C-		
ries			
	C-1 1/2		
	Card 1/2	24-4-31	
2011 20 20 20 20 20 10 PM		erozen abera	versus son acrises

L 60045-65

ACCESSION NR: AP5018036

The synthesis of such silicon containing polyester amides having a space-lattice structure was studied by condensing pentaerythritol tetrasebacate, pentaerythritol tetraadipate, and pentaerythritol tetrasuccinate with bis(phenylaminomethyl) tetramethyldisiloxane and bis(phenylaminomethyl)hexadecamethyloctasiloxane. It was found that the rate of polycondensation of the tetramethyldisiloxane with the tetrafunctional ester acids up to the gel point of the polymers and also after the start of gelling decreases with increasing distance between the carboxyl groups of the tetrafunctional ester acids. Condensation of adipic acid with bis (phenylaminomethyl) tetramethyldisiloxane produced linear polyamides. The reaction, carried out at 160C, is not accompanied by degradation of the components. The new compounds &, W-bis(chloromethyl)hexadecamethyloctasiloxane and d, urbis(phenylaminomethyl)hexadecamethyloctasiloxane were synthesized. Condensation of the latter compound with tetrafunctional ester acids at 160C occurs in a heterogeneous medium and yields inhomogeneous products forming fibers. Original art. has: 4 figures, 3 formulas and 1 table.

ASSOCIATION: None

SUBMITTED: 00

ENCL:

SUB CODE: OC

NO REF SOV: 004

OTHER: 002

ANDRIANOV, K.A.; YEMEL'YANOV, V.N.

Three-dimensional condensation of silicon-containing alcohols with tetrafunctional acid esters. Vysokom. soed. 7 no.3:517-522 Mr '65. (MIRA 18:7)

1. Institut elementoorganicheskikh soyedineniy AN SSSR.

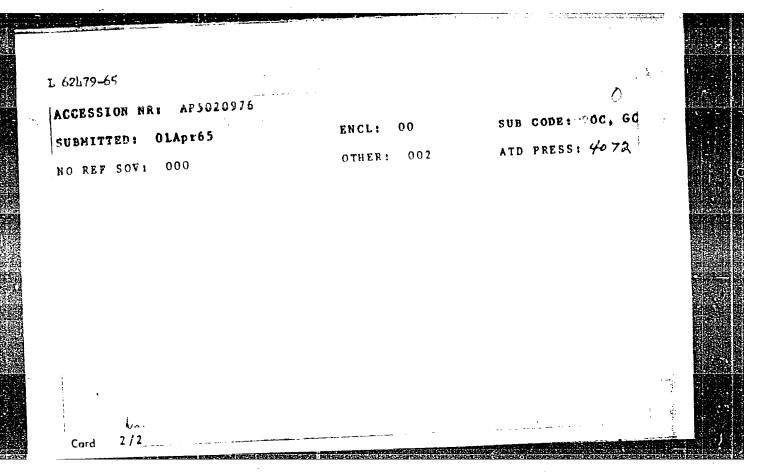
ANDRIANOV, K.A.; GOLUBKOV, G.Ye.; YELINEK, V.I.; KURASHEVA, N.A.; MANUCHAROVA, I.F.; LITVINOVA, L.F.; ARTEM YEV, B.K.

Synthesis and properties of polytitanodimethylsiloxanes. Vysokom. soed. 7 no.4:680-687 Ap '65.

X-ray diffraction study of polytitanodimethylsiloxanes, Ibid.: 688-695 (MIRA 18:6)

1. Institut elementoorganicheskikh soyedineniy AN SSSR, Institut obshchey i neorganicheskoy khimii imeni Kurnakova i Vsesoyuznyy elektrotekhnicheskiy institut imeni Lenina.

1	62479-65 ENT(m)/EPF(c)/ENP(j)/T RM	i.
_	United to 100 1 / 100 7 / 100 7 / 1 / 1 / 1 / 1 / 1 / 1 / 1 / 1 / 1 /	
¥	A CLESS LOBERT NEW YORK OF THE STATE OF THE	
A	AUTHOR: Andrianov, K. A.; Kurakov, G. A.; Sustents va, F. F.; Myagkov, V. A.; Avilov, V. A. M.	
<u>,</u>	MISS AVITOV, V. A. 44 55	
-	TITLE: Polymerization of cyclic phenylailseaquioxanes	
	SOURCE: Vysokomolekulysrnyye soyedineniya, v. 7, no. 8, 1965, 1477	
•	TOPIC TAGS: organosilicon polymer, silicone, phenylsilsesquioxane	
- 1	ABSTRACT: High-molecular-weight, benzene-soluble polymers having a	
	the transfer of above 1000 three pech property	
<u>`</u> 4∏1	the cyclic phenyisilsesquioxane octamer (busiless the octamer were unsuccessful,	
	an alkali to complete dissolution of the starting material	
•	without solvent at 250-270C.	
•	ASSOCIATION: none	
:	NOUVOLANT ACTION OF THE PROPERTY OF THE PROPER	
	Cord 1/2	



ANDRIANOV, K.A.; MANEVICH, I.Ya.; BUSLAYEV, Yu.A.; MATROS)V, Ye.I.

Acid salts of methylphosphiric acid. Zhur. neorg. khim. 10 no.3:596-600 Mr '65. (MIRA 18:7)

1. Institut elementoorganicheskikh soyedineniy AN SSSR i Institut obshchey i neorganicheskoy khimii imeni N.S. Kurnakova AN SSSR.

AMBRIANOV, K.A., akademik; YEEELTYANOV, V.N.; SUKHAREVA, L.A.; SETERIOVA, Yu.P.;
ZUBOV, P.I.

Synthesis and physicomechanical properties of films of polymers
of regular structure. Dokl. All SSSR lol no.1:99-102 Mr 65.

(MIRA 18:3)

1. Institut elementoorganicheskikh solvedineniy AN SSSR.

ANDRIANOV, K.A., akademik; PAKHOMOV, V.1.; GEL'PERINA, V.M.

Disproportionation reaction used as a new method for synthesizing organosilicon polymers. Synthesis of polyphenylenesilanes. Dokl. AN SSSR 162 no.1:79-81 My '65. (MIRA 18:5)

1. Nauchno-issledovatel'skiy institut plasticheskikh mass.

ZHDANOV, A.A.; ANDRIANOV, K.A., akademik; ODINETS, V.A.; KARPOVA, I.V.

Synthesis and polymerization of cyclotetrasiloxanes containing heterocyclic radicals with a silicon atom. Dokl. AN SSSR 162 no.21335-338 My 165. (MIRA 18:5)

1. Institut elementoorganicheskikh soyedineniy AN SSSR.

ANDRIANOV, K.A., CRAIDEMERY KOLESNIKOV, G.S., RODIONOVA, Ye.F.: LUKIYANOVA, G.M., PERTSOVA, N.V.

Thermal degridation of the polymers of vinylpheaplinic acid esters. Dokl. AN SSSR 163 no.1:97-99 J1 '65. (MIRA 18:7)

1. Institut elementoorganicheskikh soyedineniy AN SSSR i Moskovskiy khimiko-tokhnologicheskiy institut im. D.I. Mondeleyeva.

ANDRIANCY, E.A., Ekademik; FEDIN, E.I.; KOTRELEV, G.V.; GORSEAYA, I.V.

High-resolution proton magnetic resonance of organocyclosilazanes.
Dokl. AN SSSR 163 no.4:877-879 Ag *65. (MIRA 18:8)

1. Institut elementoorganicheskikh soyedineniy AN SSSR.

EWI(m)/EPF(c)/EMP(j)/T JAJ/RM Pc=4/Pr-4 UR/0062/65/000/006/1022/1025 ACCESSION NR: AP5017960 542.91 Andrianov, K. A.; Tsomaya, N. I.; Khananashvili, L. M. Synthesis of glycerol derivatives of oligomers with terminal functional TITLE: groups SOURCE: AN SSSR. Izvestiya. Seriya khimicheskaya, no. 6, 1965, 1022-1025 TOPIC TAGS: glycerol, glycol, hexamethylenediamine, oligomer, adipic acid, sebacic acid ABSTRACT: The development of methods for preparing polymers vis oligomers involves the development of a synthesis of oligomers with terminal reactive groups. The article presents data on the synthesis of esters from glycerol and various dibasic acids, as well as data on the synthesis of oligoners from the glycerides of dibasic acids and hexamethylenediamine or a glycol. A study of the reaction between glycerol and dibasic acids showed that the reaction must be carried out in the presence of a large excess of acid to permit the formation of tribasic glyceroscids. The reaction follows the equation CH₂O-CO (Cit₂)_n COOH CHIOH 1/3 Card

`*		H•0H + 3H00C (CH ₂) _n C	"(вНЭ) ОЭ—ОНЁ) ← НООС (вНЭ) ОЭ—ОвНЁ	('I	
	n = 4.8	(11gO11	Chips-ado (ding)	,,00011	
The oligomers wer methylenediamine	re synthesized by and glycols. It acids to the other	condensing tribat was found that a component is life	us obtained and id sic glyceroacids w lso in this case, 5, the reaction pr	when the	
the formation or	Oligonicio do 201	.0			, t
	0 (CH ₁) _n СООН) _n CONH (CH3)¢NH3	1 20	: : 2: :.
t) CH ₂ OCO	9 (СН ₁) ₂ СООН	CH30C0 (CH3),,CONH (CH3);NH3 ,, CONH (CH3); NH4 + 3H;	,	\$ * 3. *.
з) си _з осо	9 (СН ₁) ₂ СООН	СН•0СО (СП•), - СН•0СО (СП•),		.0	5 ⁵ 2. 5.
сн-осо сносо	9 (СН _{3)n} СООН - (СН _{3)n} СООН +3NH3 (С	сн _э осо (сн _з) - сносо (сн _з), - сносо (сн _з)	_п CONH (СП _в), NH _s + 3H	,0	\$ 1. 2. 1.
1) GH ₂ OCO CH ₂ OCO 2) GH ₂ OCO	9 (СН ₃) ₂ СООН • (СН ₃) ₂ СООН +3NH3 (С 9 (СН ₃) ₂ СООН 0 (СН ₃) ₃ СООН	СН30СО (СН3) — СН0СО (СН3), — СН0СО (СН3) — СН30СО (СН3)	$_{n}$ CONH (CH ₂), NH ₁ \pm 3H, CONH (CH ₂), NH ₃	·	6.5 2.1.
1) СИ ₁ 0С0 СИ ₂ 0С0 2) СИ ₁ 0С0 СИ ₂ 0С0 СИ ₂ 0С0	9 (СН ₃) ₂ СООН • (СН ₃) ₂ СООН +3NH3 (С 9 (СН ₃) ₂ СООН 0 (СН ₃) ₃ СООН	сносо (си _з) сн _з осо (си _з), сн _з осо (си _з) сн _з осо (си _з)	_л CONH (CH ₃) ₉ NH ₄ + 3H ₅) _в CONH (CH ₃) ₉ NH ₉ ₃) _в COO (CH ₃) ₈₈ OH	·	6.3 2. 1,

L 60047-65

ACCESSION NR: AP5017960

These oligomers were separated by removing excess hexamethylenediamine or glycol by extraction with water. The properties of the synthesized oligomers are given, and the experimental procedure followed for each compound is described. Orig. art. has: 2 tables and 3 formulas.

ASSOCIATION: Moskovskiy institut tonkoy khimicheskoy tekhnologii im. M. V. Lomonosova (Moscow Institute of Fine Chemical Technology)

SUEMITTED: 20May 63

ENCL: 00

SUB CODE: OC

NO REF SOV: 000

OTHER: 000

Card

ACCESSION NR: AP5018911	UR/0363/65/001/006/0825/0829 3'/ 541.64 35
AUTHOR: Andrianov, K. A.; Yer	makova, M. N. 55
TITLE: Polymers with inorganic	molecular chains - polyboromethylphenylaminomethyl-
siloxanes q, 55	
SOURCE: AN SSSR. Izvestiya. No	eorganicheskiye materialy, v. 1, no. 6, 1965, 825-829
TOPIC TAGS: boron organic com	pound, polysiioxane, silicon organic polymer
branched structure, an attempt we gomers by introducing into the incredicals having donor properties.	mation of polyborodimethylsiloxane oligomers with a as made to raise the hydrolytic stability of these oligographic chain siloxane links surrounded by organic. To this end, use was made of the phenylaminomethyltroduced into the molecules in small amounts by condensiloxanes with phenylaminomethylmethyldiethoxysilane:
R R	R 11 R 1 1 1 1 1 1 1 1 1 1
$H(OSi)_{n}-OH+C_{2}H_{5}O-S$	$i - OC_0H_0 \rightarrow 2C_0H_0OH + C_0H_0O - Si - (OSI)_n - OSI - OC_0H_0$ (1)

L 60953-65

ACCESSION NR: AP5018911

where $R = CH_2$ and $R' = -CH_2NHC_6H_5$. The reaction, associated with a quantitative evolution of ethyl alcohol, was used to control both the number of methylphenylaminomethylsiloxane groups and their position relative to the siloxane bond. The reaction of polycondensation of α , W-dihydroxydimethylsiloxanes with phenylaminomethyldiethoxy-silane was studied with oligomers of various degrees of polymerization. Oligomers of the general formula

were obtained, where $R = CH_3$ and $R' = CH_2NHC_6H_5$, with functional ethoxy groups at the ends of the inorganic chain of molecules, and n = 14, 25, 46, 74, and 188. The properties of these oligomers were determined. Branched borophenylaminomethylmethylsiloxanes with a regular distribution of boron in the inorganic chain were prepared by condensing the synthesized oligomers with boric acid. The kinetics of the cross-linking reaction of the polymer obtained from the various oligomers were studied. The rate constant of this reaction decreases with the degree of polymerization of the oligomer.

Cord 2/3

	ACCESSION NL AP5018911 Orig. art. has: 3 figures, 2 table			2
Elektria	ASSOCIATION: Institut elementos stitute of Organometallic Compou	rranichoskikh sovedinaniv	Akademii nauk SSS	R (In-
+	SUBMITTED: 17Feb65	ENC L: 00	SUB CODE: C	oc, ic
	NO REF SOV: 602	OTHER: 001		
		·		
	dm Cord 3/3			

EWT(m)/EPF(n)/EWP(j)/T Pc-4/Pr-4UR/0190/65/007/004/0680/0687 <u>l. 51393-65</u> ACCESSION NR: AP5011249 AUTHORS: Andrianov, K. A.; Golubkov, C. Ye.; Yelinek, V. I.; Kurasheva, N. A.; Manucharova, I. F.; Litvirova, L. F.; Artem'yev, E. K. TITLE: Synthesis and properties of polytitanodimethylsiloxanes SOURCE: Vysokomolekulyarnyye soyedineniya, v. 7, no. 4, 1965, 680-687 TOPIC TAGS: organic synthesis, thermographic analysis, glass transition temperature, polycondensation The authors synthesized titanodimethylsiloxane oligomers of the general ABSTRACT: formula Ti(OSi-).OH], in which n may equal 9, 18, 25, 34, 42, 52, 80, or 104. Polycondensation was carried out at 2000. Thermographic analyses were made of the titancdimethylsiloxane oligomers and polymers, and the heat of fusion of the crystalline phase was determined; these data are tabulated. Electrical studies show that an increase in content of the hydroxyl group in the titanodimethylsiloxane oligomers leads to an increase in rigidity, a retardation in orystallization, and an increase in polarizability and dielectric loss. It is concluded that the relaxation

	L 51393-65 ACCESSION NR: AP5011249	programme and the second of th	3
00	reaction of the compounds amorphous phase. The time of the polydimethylsiloxan found to lie within the ratables.	in an electric field is determined by of relaxation was found to be indepe e branches. The glass point for the nge from -120 to -1300. Orig. art. h	studied compounds was cas: 2 figures v.3
	Hetero-Organic Compounds	An SSH); Institut Obsidery I heorga-	Vaesoyuznyy
	Hetero-Organic Compounds	of General and Inorganic Chemistry); tut in . V. I. Lenina (All-Union Elect	Vaesnyuznyy
121	Retero-Organic Compounds, N. S. Kurnakova (Institute elektrotekhnicheskiy insti Instituts)	of General and Inorganic Chemistry); tut in . V. I. Lenina (All-Union Elect	Vaesoyuznyy
	N. S. Kurnakova (Institute elektrotekhnicheskiy insti	of General and Inorganic Chemistry); tut is 7. I. Lenina All-Union Elec-	Vaesnyuznyy
	Retero-Organic Compounds, N. S. Kurnakova (Institute elektrotekhnicheskiy insti Institute) SURMITTED: 27Jun64	of General and Inorganic Chemistry); tut i. 7. I. Lenina All-Unin Elect	Vaesnyuznyy
	Retero-Organic Compounds, N. S. Kurnakova (Institute elektrotekhnicheskiy insti Institute) SURMITTED: 27Jun64	of General and Inorganic Chemistry); tut i. 7. I. Lenina All-Unin Elect	Vaesnyuznyy

<u>L 51394-65</u> EWT(m)/EPF(c)/EWP(j)/T Pc-4/Fr-4 RM	
ACCESSION NR: AP5011250 UR/0190/65/007/004/0688/0695	
AUTHORS: Andrianov, K. A.; Golubkov, G. Ye.; Yelinek, V. I.; Kurasheva, N. A.; 30 Manucharova, I. F.; Litvinova, L. P.; Artem yev, B. K.	B
TITLE: X-ray studies of polytitanodimethylsiloxanes	
SOURCE: Vysokomolekulyarnyye soyedineniya, v. 7, no. 4, 1965, 688-695, and insert facing p. 688	
TOPIC TAGS: x ray study, polymer, titanium compound, structure analysis	
ABSTRACT: The structure of polytitanodimethylsiloxanes was studied by means of x-ray structure analysis at room temperature and at -1201. The x-ray photographs were taken with CuK radiation (>= 1.54 Å), and the interplanar distances were computed by the Bragg-Wulff formula. The data are tabulated. Results show that the	
oligomera and polymera are amon house at room temperature. The structure is characterized by bundle packing of the dimethylsilozane branches, identical to the packing	
of molecular chains in polymethylsiloxane. It was found that titanodimethylsiloxane oligomers with terminal hydroxyl groups for $n \ge 42$, polymers with $n \ge 36$, and oligomers with terminal trimethylsilane groups with $n \ge 23$ orystallize in approximatel, the same temperature range as polydimethylsiloxanes, and the structures of	
Cord 1/2	

51394-65 CCESSION NR: AP5011250	* ***	3
	entical. Orig. art. has: 4 fig	~ 1
atera-Organia Compounda, AN	toorganicheskikh soyedineniy Al SSSR); Institut obshchey i neo Ceneral end inorganic hem so im. V. I. bedina Allenion E	TYANTONGGRIKH KILLETI III.
UBMITTED: 27Jun64	ENCL: CO	SUB CODE: OC, OP
0 REF SOV: 002	OTHER: 000	

L 2267-66 EWT(m)/EPF(c)/EWP(j) RM ACCESSION NR: AP5022223 UR/0191/65/000/009/0020/0022 Andrianov, K. A.; Pakhomov, V. I.; TITIE: Reactions of chloromethylsilanes and siloxanes with dihydric phenols SOURCE: Plasticheskiye massy, no. 9, 1965, 20-22 TOPIC TAGS: organosilicon compound, resorcinol, hydroquinone, condensation reaction, silane esterification ABSTRACT: The object of the work was to synthesize monomeric hydroxyphenoxymethylsiloxanes and monomeric products with completely esterified phenol hydroxyls by the reaction of chloromethylalkoxysilanes and siloxanes with monoand diosodium derivatives of dihydric phenols. As a result of the reaction, the chlorine in the methyl radical was replaced by the residue of the dihydric phenol containing a free or esterified hydroxyl group. Reactions of bis(chloromethyl) tetramethyldisiloxane and chloromethyldimethylbutoxysilane with resorcinol, hydroquinone, or dihydroxydiphenylpropane (diane) were carried out in absolute butanol under nitrogen. The conditions and results of these reactions are tabulated. The silanols obtained were subjected to condensation reactions, and 1/2 Card

in the second		
	L 2267-66 ACCESSION NR: AP5022223)
•	the transesterification of methylphenyldiethoxysilane and phenyltriethoxysilane was performed. Orig. art. has: 3 figures and 1 table.	
	ASSOCIATION: none	
	SUBMITTED: 00 ENCL: 00 SUB CODE: OC	
1	NO REF SOV: 001 OTNER: 003	
	Card 2/2 (

L 2168-66 EWT(m)/EPF(c)/EWP(j) RM UR/0191/65/000/010/0015/0016 ACCESSION NR: AP5024501 678. 84.4 AUTHOR: Semenova, Ye. A.; Makovskaya, T. N. Zhinkin, D. K. A. 4455 TITLE: Rearrangements of methylcyclosilazanes 1,44,55 SOURCE: Plasticheskiye massy, no. 10, 1965, 15-16 TOPIC TAGS: organosilicon compound, chemical reaction, chemical equilibrium recombination reaction, chemical reaction kinetics, catalytic polymerization, catalysis ABSTRACT: The effect of electrophilic catalysts at different temperatures on the mutual rearrangements of methylcyclosilazanes was investigated to explain previously obtained data. The conversion of hexamethylcyclotrisilazane (A) and octamethylcyclotetrasilazane (B) by the action of 2% ammonium sulfate of 1% concentrated sulfuric acid was studied in the 25-245 C range. Mutual rearrangement of the two cyclosilazanes occurred, and at temperatures above 150 C polymethylsilazanes were formed. The latter polymers were viscous yellow liquids having a cyclo-linear structure. Ammonium sulfate was most active in the

	L 2168-66		·			Ē
	ACCESSION NR: AP5024501		i ili ili ili ili ili ili ili ili ili i	ing to the mean more and the control of the control	0	
	ring-contracting reaction and prome above 100 C. The composition of the acid on A or B at a given temperatu figures	e products obt	ained by actio	n of sulfurio		1
	ASSOCIATION: None					
	SUBMITTED: 00	ENCL: 00	SUB CO	DE: OC, L	12	
	NR REF SOV: 003	OTHER: 000				
•						
						0
	Card 2/2/)					
,	Cara 27 20	an disemply property and a		يارد المدويسية ومستسمهمية ب		1

ANDRIANOV, K.A.; KUZNETSOVA, I.K.; SMIRNOV, Yu.N.

Reaction of tetrabutyl titanate with methylphosphinic acid esters. Izv. AN SSSR. Neorg. mat. 1 no.3:289-293 Mr '65.

Reactions of n-tetrabutyl titanate with phosphinic acids.

Ibid.:301-306 (MIRA 18:6)

1. Institut elementoorganicheskikh soyedineniy AN SSSR.

ANDRIAMOV, K.A.; KURASHEVA, N.A.; MANUCHAROVA, I.F.; MILLINER, Ye.M.

Polymers with inorganic chains of molecules ...
polytitanodimethylailoxanes. Izv. AN SSSR. Neorg. mat. 1
no.3:294-300 Mr 165. (MIRA 18:6)

1. Institut elementoorganicheskikh soyedineniy AN SSSR i Institut obshchey i neorganicheskoy khimii imeni Kurnakova AN SSSR.

ANDRIANOV, E.A., YERMAKOVA, M.N.

Polyboronmethylphenylaminomethylsiloxanes, polymers with inorganic molecular chains. Izv. AN SSSR. Neorg. mat. 1 no.6:825-829 Je '65. (MIRA 18:8)

1. Institut elementoorganicheskikh soyedineniy AN SSSR.

TROPINGVA, I.V.; ANDREADEL, R. J. D. TRUME, M.A.; RELIVERED, D.A.

Synthesis of mathylabilorenilese in a Flatelized ted by means of vibration. Knim.prom. All no.6:468-470 Se *165.

(MRA 18:8)

ANDRIANOV, K.A.; OVCHINNIKOV, V.A.; KHANANASHVILI, L.M.

Organosilicon heat carriers. Khim. prom. 41 no.8:570-572 Ag *65. (MIRA 18:9)

ACCESSION NR: AP5016503	UR/01-0/65/007/006/1000/1004
AUTHORS: Andrianov, K. A.; Lavygin, I.	3.2 3.0 3.0
TITLE: Formation of three-dimensional ane polymers	8-hydroxyquinolinetitanopolydimethylailox-
SOURCE: Vysokomolekulyarnyye soyedinen	iya, v. 7, no. 6, 1965, 1000-1004
TOPIC TAGS: silomane, organosilicon co tion metal complex, reaction mechanism	mpound, polymer, resin, oligomer, transi-
ABSTRACT: The work was initiated to el merization. The investigation is an ex	ucidate the mechanism of oligomer poly- tension of the work of K. A. Adrianov and
A. A. Zhdanov (Izv. AN SSSR. Otd. khim.	n., 1962 837). The oligomers studied Si(CH ₃) ₂ O)nH ₃ (I). The polymerization of
these substances with themselves and wi (II) was investigated at 2000 in vacuum and II was of first and second order, r	th 8-hydroxyquinolinetributoxytitanium . It was found that polymerization of I

L 60137-65

ACCESSION NR: AP5016503

thermodynamic properties for the systems studied are given as functions of condensation time. A reaction mechanism is proposed. Orig. art. has: 1 table, 5 graphs, and 2 illustrations.

ASSOCIATION: Institut elementoorganicheskikh soyedineniy AN SSSR (Institute for Hetero-Organic Compounds, AN SSSR)

SUBMITTED: 07Jul64

ENCL: 00

SUB CODE: OC

NO REF 30V: 005

CTHER: 000

Card 2/2

L 1128-66 EWT(m)/EPF(c)/EWP(j) RPL WW/RM		
ACCESSION NR: AP5022931 UR/0062/65/000/008/13 4μ/55 AUTHOR: Tkeshelashvili, R. Sh.; Andrianov, K. A.; Nogaydeli, A. I.	96/1402 5 2 4 2 1	
TITLE: Reaction of dimethyl- and phenylmethyldichlorosilanes with 1,4-di:	Lithium-	
SOURCE: AN SSSR. Izvestiya. Seriya khimicheskaya, no. 8, 1965, 1396-1402		
TOPIC TAGS: dimethyldichlorosilane, condensation reaction		
ABSTRACT: The reaction of dimethyl- and phenylmethylchlorosilanes with diderivatives of naphthane was studied to determine its usefulness in the sy of oligomers. The condensation reaction of 1,4-dilithium-1,4-dihydronaphy with dimethylchlorosilane proceeds according to the following scheme	nthesis	
H LI CH ₈ H CH ₉ CH ₈ CH ₉ CH ₈ CH ₉ CH ₈ CH ₈ CH ₉ CH ₈ CH ₈ CH ₉ C	CH _s	
Card1/3		
	The same of the sa	

L 1128-66

ACCESSION NR: AP5022931

The reaction product is a tetramer with a boiling temperature of 218-220°C (at 1 mm Hg). In the absence of moisture this reaction proceeds according to

This scheme was followed also in the case of condensation with phenylmethyldichlorosilane. In this case the products were: a dimer boiling at 200-205°C (1 mm Hg) and a tetramer boiling at 245-250°C (1 mm Hg). Boiling temperatures at reduced pressure, refractive indices, and molecular weights (elemental analysis) were determined for all reaction products. In order to confirm the structure, the reaction products were hydrolyzed to the corresponding dihydroxy-derivatives with various degrees of

Card 2/3

L 1128-66	and the second second	a. Programme desire day gampage yep gage									
ACCESSION NR	: AP5022931	L						•	•	خ	
polymerizati	on and trans	sformed into	other	derivat	ives.	Orig.	art.	has:	2 ta	bles.	
ASSOCIATION: (Institute o	Institut e F Elemental	elementourge Organic Co	anichesk moounds.	1kh soy Acader	redinen nv of S	ily Akad Science:	demii SSS)	nauk R)	SSSR		
						5					
SUBMITTED:	09Ju164		ENCL:	00			SUB	CODE	GC,	oc	13.1
NO REF SOV:	001		OTHER:	000							
							1:				
	•										
			•				•				10 m 2
•											
Card 3/3											1 3 3 5 1 3 3 4
	·	***************************************	·						 ,		

I. 00752-66 EPA(s)-2/IMT(m)/EPF(c)/EAP(1)/T RM

ACCESSION NR: AP5020974 UR/0190/65/007/008/1456/1462

AUTHOR: Andrianov, K. A.; Fromberg, M. B.; Belkina, T. M.

AUTHOR: Andrianov, K. A.; Fromberg, M. D., Berkha, T. M.

TITLE: Synthesis of trifunctional crosslike ester acids and of polyesters having a regular lattice structure

SOURCE: Vysokomolekulyarnyye soyedineniya, v. 7, no. 8, 1965, 1456-1462

TOPIC TAGS: ester, polyester plastic, polycondensation, adipic acid, dielectric loss, synthesis, polymer structure

ABSTRACT: Trifunctional ester acids were synthesized by reacting trimethylolethane or trimethylolpropane with a two-fold excess of adipic, azelaic or sebacic acid. Their properties were determined. The kinetics of the polycondensation of trimethylolethane and adipic acid in 1:1.5 and 1:6 ratios were investigated. In the first case the reaction is of the second order and in the case with excess adipic acid the reaction is first order. The reaction rate constants and the energy of activation of these polycondensations were calculated: E= 20,600 cal/mol

Card 1/2

APPROVED FOR RELEASE: 03/20/2001

CIA-RDP86-00513R000101420005-8"

ANDRIANOV, K.A.; KURAKOV, G.A. SUSHEMISOVA, T.F.; MYAGKOV, V.A.; AVILOV, V.A.

Polymerization of cyclic phenylsilsesquioxanes. Vysokom. soed.
7 no.8:1477 Ag 165. (MIRA 18:9)

ANDRIANOV, K.A.; KURASHEVA, N.A.; AVILOV, V.A.

Condensation of A, Q-dihydroxydimethylsiloxanes with tetrabutoxytitanium. Izv. AN SSSR Ser. khim. no.9:1616-1619 165. (MIRA 18:9)

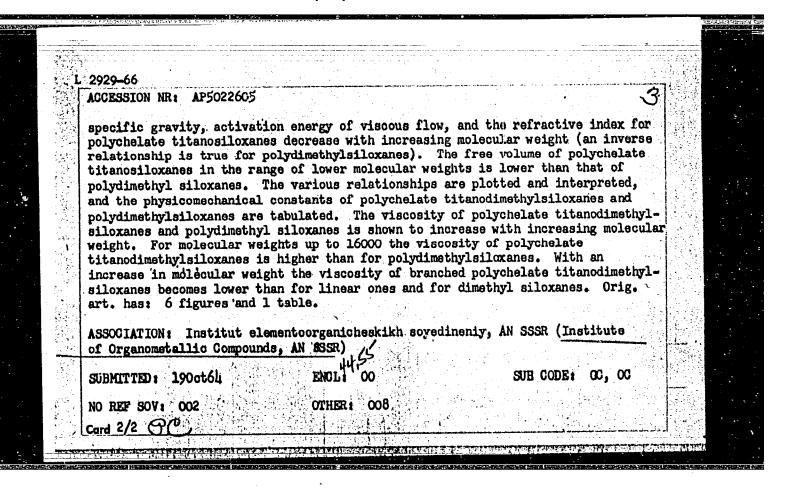
1. Institut elementoorganicheskikh soyedineniy AN SSSR.

PFTRAShilo, Yu.K.; FROMBERG, M.R.; ANDRIANOV, K.A.

Synthesis of ramified oligomers with the central silicon atom and hydroxyl groups at the ends of branching. Izv. AN SSSR. Ser. khim. no.9:1709-1711 *65. (MIRA 18:9)

1. Elektrotekhnicheskiy institut im. V.I. Lenina.

	L 2929-66 EWT(m)/EPF(c)/EWP(j)/T RM	() 上 () () () () () () () () () () () () () (
	ACCESSION NR: AP5022605 UR/0190/65/007/009/1585/1591 678.01:53+678.84	
	AUTHORS: Andrianov, K. A.; Lavygin, 1. A.	
116	TITLE: The structure and properties of linear and branched polychelate titanodimethylsiloxanes 1 44.5	
	SOURCE: Vysokomolekulyarnyye soyedineniya, v. 7, no. 9, 1965, 1585-1591	
	TOPIC TAGS: linear polymer, branched polymer, dimethylsiloxane, titanium	
	ABSTRACT: The physicochemical properties of two polymerohomologous series of linear and branched polychelate titanosiloxanes of the general formulas (C9H6NO) ₂ Ti- \(\overline{O}(Si(CH ₃) ₂ O) _n Si)(CH ₃) ₃ . \(7_2\) and (C9H6NO) Ti \(\overline{O}(Si(CH ₃) ₂ O) _n Si(CH ₃) ₃ . \(7_3\) were investigated and their properties were compared with those of linear	
	polydimethylsiloxanes. It was found that the introduction of titalium atoms and appropriate the siloxane chain increases	
	the molecular interaction and influences the physicochemical properties of the investigated compounds. Specific gravity and activation energy were found to vary	
	weight and temperature was noted for both compounds. It was demonstrated that the Card 1/2	
101034004		



L 1255-66 EPF(c)/EWP(j)/EWT(m)/T RM

ACCESSION NR: AP5021674 / UR/0080/65/038/008/1887/1889

405 547. 211'222'245

AUTHOR: Lobusevich, N. P.; Trofimova, I. V.; Andrianov, K. A.; Golubtsov,

TITLE: Effect of methyl chloride and vinyl chloride on the synthesis of methyl-chlorosilanes 1 345

SQURCE: Zhurdal prikladnoy khimii, v. 38, no. 8, 1965, 1887-1889

TOPIC TAGS: chloride, silane, methylene chloride, vinyl chloride, catalysis, copper, silicon, aluminum, antimony

ABSTRACT: Methyl chloride obtained by chlorination of natural gas contains up to 1.7 vol.% methylene chloride and 0.2-3.0% vinyl chloride. It is known that at temperatures of 300-350C methylene chloride reacts with silicon copper catalysts with formation of hexachlorodisilane methane and also of hydrogen containing chlorosilanes. Under the conditions of the reaction of methyl chloride with silicon-copper catalysts, the methylene chloride can react with the silicon with information of analogous compounds, and can undergo decomposition with the formation of carbon, which deactivates the catalyst. Carbonization of the catalyst was observed even after short term synthesis, with the introduction of Cord 1/2

L 1255-66

ACCESSION NR: AP5021674

6-7% of methylene chloride into the methyl chloride. In experiments in a pressur ized fluidized bed on an alloy promoted with antimony, an investigation was made of the effect of vinyl chloride, whose concentration in the mixture with methyl chloride was varied from 0.16 to 4.0 vol. %. No adverse effect on the process was observed at concentrations up to 0.2%. In the reaction of methyl chloride with an alloy of the composition Cu3Si, vinyl chloride in concentrations higher than 0. 16% sharply lowers overall activity and slightly lowers selective activity. For Cu₃Si alloys and mixtures of copper and silicon powders with addition of 0.5% aluminum, the introduction of more than 0.16% vinyl chloride causes a greater decrease in overall activity than for catalysts with an antimony additive. In this case, large amounts of still residues are formed (15-40%). In general, it is concluded that under the conditions of the synthesis, vinyl chloride reacts with silicon with the formation of vinyl trichlorosilane, ethyl dichlorosilane, and dimethyl vinyl chlorosilane, and that this inhibits the separation of dimethylchlorosilane from the mixture of methylchlorosilanes. Orig. art. has: 3 figures and 1 table

SUBMITTED: 17Jun63

NR REF SOV: 003

Card 2/2 KC

ENCL: 00 OTHER: 002

SUB CODE: MM, GC

4535-00 Las. (E., -11, C., 141) (J) ACC NR. AP5027692 SOURCE CODE: UR/0062/65/000/010/1895/1897 Andrianov, K. A.; Lavygin, I. A AUTHOR: ORG: Institute of Heteroorganic Compounds, Academy of Sciences, SSSR (Institut elementoorganicheskikh soyedineniy, Akademiya nauk SSSR) TITLE: The reaction of bis-(8-hydroxyquinoline)dibutoxytitanium with a-hydroxy-w-(trimethylsilyl)dimethylsiloxanes SOURCE: AN SSSR. Izvestiya. Seriya khimicheskaya, no. 10, 1965, 1895-1897 TOPIC TAGS: titanium compound, titanium organic compound, siloxane, titanosiloxane, 8 hydroxyquinoline ABSTRACT: The condensation of bis(8-hydroxyquinoly1)dibutoxytitanium with \u03c4-hydroxyw-(trimethylsilyl)dimethylsiloxane resulted in the formation of linear oligomers with the structure: where n was 15, 60, 98, 170, or 350. The condensation was performed in benzene solution at 80C for 3-4 hours. Butanol was split off. The oligomers obtained were vis-**Card** 1/2 09010027

